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3,1-Benzoxazin-2-ones and 3,1-Benzoxazine-2-thiones

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3-exo-Aminobicyclo[2.2.1]hept-5-ene-2-exo-carboxylic acid and ethyl 3-endo-aminobicyclo[2.2.1]heptane-2-endo-carboxylate (6) were reduced with lithium aluminum hydride to the corresponding bicyclic aminoalcohols 3 and 4. These and the saturated endo-endo and exo-exo N-methylaminoalcohols 16 and 22, respectively, were converted to methylene-bridged tetrahydro- (11) and hexahydro-3,1-benzoxazin-2-ones 12, 17, 23 and 3,1-benzoxazin-2-thiones 13, 14, 18, 24. The exo-exo 3 and endo-endo 4 aminoalcohols were cyclized by means of ethyl arylimidates to tricyclic dihydro-1,3-oxazines 7a-d, 8a-d. The structures were confirmed by ir, ¹H and ¹³C nmr spectroscopy.

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We earlier reported the synthesis of the 3-endo-aminobicyclo[2.2.1]hept-5-ene-2-endo-carboxylic acid (1) and the isomeric saturated exo amino acid 2, and the reduction of these to 1,3-aminoalcohols [3], from which, by cyclization, the corresponding 1,3-oxazines, 1,3-oxazin-2-ones and 1,3oxazine-2-thiones fused with norbornene or norbornane were prepared. Starting from the 1,3-oxazines, we prepared by cycloaddition isomeric azetidinones, the structure and steric structure of which were elucidated by 'H and ¹³C nmr spectroscopy [4]. That work was carried out for pharmalogical purposes and to make systematic comparative stereochemical studies. As a continuation, we endeavoured to prepare further stereoisomeric 1,3-oxazines, 1,3-oxazin-2-ones and 1,3-oxazine-2-thiones fused with the norbornane skeleton. Whereas in the compounds prepared earlier the oxygen of the oxazine ring is attached to the carbocyclic unit [5], in the isomers described in the present work the nitrogen is the linking atoms.

By acylation and cyclization of the *endo* amino acid 1, tricyclic 1,3-oxazin-4-ones fused with norbornene were prepared which by retrodiene decomposition on pyrolysis furnished 6*H*-1,3-oxazin-6-ones [6], compounds difficult to obtain by other means. This reaction was also applied for the *exo*-1,3-oxazin-4-ones, and the rates of decomposition of the isomers were studied kinetically [7].

In the present work we report the preparation of further isomeric exo-exo 1,3-oxazines and related compounds containing norbornene as structural unit. An account is given of the saturated carbocyclic derivatives of the endo-endo norbornene compounds described earlier [3] and of some N-methyl-substituted 1,3-oxazin-2-ones and 1,3-oxazine-2-thiones.

Synthesis.

The starting aminoalcohol 3 was prepared from 3-exo-aminobicyclo[2.2.1]hept-1-ene-2-exo-carboxylic acid [7] by reduction with lithium aluminum hydride (Scheme 1). For synthesis of the bicyclic saturated aminoalcohol 4. The endo-amino acid 1 [3] was esterified, the double bond of the ester salt 5 was hydrogenated, and the base liberated from the salt 6 was reduced with lithium aluminum hydride.

Starting from the bicyclic aminoalcohols 3 and 4, the 5,6-methano-r-4a,c-5,c-8,c-8a-tetrahydro-4H-3,1-benzoxazines 7a-d and the 5,8-methano-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazines 8a-d were prepared with ethylarylimidates (Scheme 2).

With ethyl chloroformate, the aminoalcohols 3 and 4 afforded the carbamates 9 and 10, which were cyclized with sodium methoxide to the 5,8-methano-r-4a,c-5c-8,c-8a-tetrahydro-4H-3,1-benzoxazin-2(1H)-one (11) and the 5,8-methano-1-methyl-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazin-2(1H)-one (12), respectively. The corresponding thiones 13 and 14 were synthesized from the aminoalcohols 3 and 4 by reaction with carbon disulphide and cyclization of the resulting non-isolated dithiocarbamates with lead(II) nitrate.

Ar = a, C_6H_5 b, $C_6H_4CI(m)$ c, $C_6H_4CI(p)$ d, $C_6H_4CH_3(p)$

The carbamate 15, prepared from the ester salt 6, was reduced with lithium aluminum hydride to 2-endo-methylamino-3-endo-hydroxymethylbicyclo[2.2.1]heptane (16). By the previous methods the aminoalcohols 16 yielded 5,8-methano-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazin-2(1H)-one (17) and the corresponding 2(1H)-thione 18 (Scheme 3).

For preparation of the isomeric saturated exo N-methyl analogues, the 3-exo-aminobicyclo[2.2.1]heptane-2-exo-carboxylic acid (19) [3] was esterified and the ester salt 20 was acylated to the carbamate 21. The reduction of compound 21 gave the exo methylaminoalcohol 22, which was transformed to 5,8-methano-1-methyl-r-4a,c-5,6,7,c-8,c-8a-hexahydro-4H-3,1-benzoxazin-2(1H)-one (23) and 2(1H)-thione 24 (Scheme 4).

Scheme 4

Spectroscopic Study.

The detailed spectroscopic study of the saturated compounds 7a-d, 11 and 13 and the unsaturated analogues 8a-d, 12 and 14 was reported earlier [4]. The structures of the new compounds were proved by the following spectral data.

The ir spectra of compounds 11 and 12 exhibit ν 2 \times NH and urethane carbonyl bands at 3325, 3300 and 3245, 3130 cm⁻¹ and at 1720 and 1695 cm⁻¹, respectively. The carbonyl bands of the N-methyl derivatives 17 and 23 appear at 1700 and 1690 cm⁻¹. The ν 2 \times NH bands of the thiones 13 and 14 are found at about 3180 cm⁻¹, and the intense characteristic thiocarbamate bands at 1550 cm⁻¹. The corresponding bands of the N-methyl derivatives 18 and 24 are identifiable at 1500 and 1495 cm⁻¹, respectively.

The diexo anellation of the hetero ring in compounds 7a-d and 13 is proved by the doublet splitting of the H-4 signal in the region of 3.30-3.45 ppm (J_{4,5} = 7.6-8.1 Hz). The H-4 signal of the diendo compounds 8a-d is a double doublet at about 3.90 ppm (J = 10.5 and 4.5 Hz), which in consequence of the coupling of the H-4 and NH protons is split to 8 signals at about 3.90 ppm (J = 12, 4 and 2 Hz) in

the cases of 12 and 14.

While the olefinic protons of the norbornene compounds 7a-d, 11 and 13 give signals in the interval 6.0-6.3 ppm, the overlapped (H-6,6',7,7',9,9') multiplets of the methylene groups in the norbornane analogues are observed at 1.0-1.8 ppm.

In the unsaturated compounds 7a-d, the H-5 and H-8 signals appear at about 2.6 and 3.0 ppm, but in the spectra of the saturated analogues 8a-d they are at about 2.25 and 2.6 ppm, because the —I effect of the unsaturated bond in the former compounds causes deshielding. Although the shifts change somewhat in the 2-oxo and 2-thioxo derivatives 11-14, the deshielding in the unsaturated compounds 11 and 13 is similar.

The C-2 signal in the ¹³C nmr spectra of compounds **7a-d** and **8a-d** can be characterized by the shifts at 156-158 ppm, and this is also the situation for the urethanes **11**, **12**, **17** and **23** (155.4-158.2 ppm). The thiocarbonyl carbon atom is deshielded, however, and characteristically, therefore, the C-2 signal appears in the interval 190.5-191.1 ppm for compounds **13**, **14**, **18** and **24**.

The C-6,7 signals could be assigned at 21-24 ppm in the norbornane derivatives 8a-d, and at about 138.5 and 136.5 in the norbornene compounds 7a-d. Disregarding some irregular values for the N-substituted analogues, no essential changes were experienced in this respect for the urethanes and thiourethanes.

In both the ¹H and the ¹³C nmr spectra, aromatic signals appear for compounds **7a-d** and **8a-d** and, of course, are absent from the spectra of the 2-oxo and 2-thioxo derivatives.

The C-methyl signals for compounds 7d and 8d and the

analogous signals for the N-methyl derivatives are identifiable in both the ¹H nmr (2.35 and 3.30 ppm, in 17 2.84 ppm) and the ¹³C nmr (21.3 and about 33-34 ppm) spectrum.

The above spectral data unambiguously prove the presumed structures and purity of the compounds. Use of the spectral data on the earlier-reported analogues will allow a comparison of the *exo-exo* and *endo-endo* anellated saturated and unsaturated compounds. These studies and the detailed spectroscopic data will be published later.

EXPERIMENTAL

The ir spectra were recorded in potassium bromide with Specord 75 (Jena GDR) or Bruker IFS-113v FT spectrometers. The 'H and '3'C nmr spectra were obtained in deuteriochloroform at room temperature with Bruker WH-250 or Bruker WP 80-SY spectrometers, locked on the deuterium signal of the solvent and using TMS as internal standard.

Ethyl 3-endo-Aminobicyclo[2.2.1]hept-5-ene-2-endo-carboxylate Hydrochloride (5) and Ethyl 3-exo-aminobicyclo[2.2.1]heptane-2-exo-carboxylate Hydrochloride (20).

Thionyl chloride (8 ml, 0.11 mole) was added dropwise with stirring to absolute ethanol (90 ml) at -10° . 3-endo-Aminobicyclo[2.2.1]hept-5-ene-2-endo-carboxylic acid (15.3 g, 0.1 mole) (1) [3] or 3-exo-aminobicyclo-[2.2.1]heptane-2-exo-carboxylic acid (15.5 g, 0.1 mole) (19) [3] was added in portions to the mixture, which was stirred for 30 minutes at 0°. After standing for 3 hours at room temperature, the mixture was refluxed for 1 hour and evaporated. The residue gave 16.2 g (75%) colorless crystals of 5, mp 178-180° and 13.4 g (62%) 20, mp 171-173° from ethanol.

Anal. (5) Calcd. for C₁₀H₁₆ClNO₂: C, 55.17; H, 7.41; N, 6.43. Found: C, 54.86; H, 7.54; N, 6.68.

Anal. (20) Calcd. for $C_{10}H_{18}CINO_2$: C, 54.67; H, 8.26; N, 6.38. Found: C, 54.85; H, 8.42; N, 6.22.

Ethyl 3-endo-Aminobicyclo[2.2.1]heptane-2-endo-carboxylate Hydrochloride (6).

Table I

Physical and Analytical Data on the Compounds Prepared (7a-d, 8a-d, 11-14, 17, 18, 23, 24)

Compound	Mp (°C)	Yield (%)	Formula	Calcd.			Found		
				С	H	N	С	H	N
7a	46-47	42	$C_{15}H_{15}NO$	79.97	6.71	6.22	79.91	6.54	6.36
7b	65-67	43	C ₁₅ H ₁₄ ClNO	69.36	5.43	5.39	69.18	5.20	5.31
7 c	84-86	48	C ₁₅ H ₁₄ ClNO	69.36	5.43	5.39	69.45	5.50	5.33
7d	59-61	40	$C_{16}H_{17}NO$	80.30	7.16	5.85	80.55	7.25	6.01
8a	146-148 [a]	34	$C_{15}H_{17}NO$	79.26	7.54	6.16	79.21	7.48	6.27
8b	66-67	36	C ₁₅ H ₁₆ ClNO	68.83	6.16	5.35	68.70	6.04	5.51
8c	85-87	45	C ₁₅ H ₁₆ ClNO	68.83	6.16	5.35	68.80	6.12	5.37
8d	71-73	45	$C_{16}H_{19}NO$	79.63	7.94	5.80	79.72	8.02	5.87
11	87-89	34	$C_9H_{11}NO_2$	65.43	6.71	8.48	65.24	6.53	8.44
12	93-95	32	C ₉ H ₁₃ NO ₂	64.65	7.84	8.38	64.81	7.96	8.30
13	123-125	54	C ₉ H ₁₁ NOS	59.64	6.12	7.73	59.45	6.03	7.50
14	161-163	49	C ₉ H ₁₈ NOS	58.98	7.15	7.64	59.14	7.26	7.67
17	122-124 [a]	30	$C_{10}H_{15}NO_2$	66.27	8.34	7.73	66.21	8.32	7.90
18	68-69	34	$C_{10}H_{15}NOS$	60.87	7.66	7.10	60.93	7.70	7.08
23	128-130 [a]	30	$C_{10}H_{15}NO_2$	66.27	8.34	7.73	66.40	8.43	7.87
24	98-100	33	$C_{10}H_{15}NOS$	60.87	7.66	7.10	60.71	7.58	7.23

Ester salt 5 (15.24 g, 0.07 mole) with platinum oxide catalyst (0.15 g) in ethanol (300 ml) was hydrogenated at room temperature and atmospheric pressure, with stirring. After completion of the reduction (4-6 hours) the catalyst was filtered off and the solvent was evaporated. The residue (13.9 g, 90%) gave colorless crystals from ethanol, mp 177-179°.

Anal. Calcd. for C10H18CINO: C, 54.67; H, 8.26; N, 6.38. Found: C, 54.77; H, 8.49; N, 6.20.

Ethyl 3-endo-Aminobicyclo[2.2.1]heptane-2-endo-carboxylate (6a).

Ester salt 6 (2.20 g, 0.01 mole) and triethylamine (1.0 g, 0.01 mole) in acetone (20 ml) was chilled to 10° for 10 minutes and the solid was filtered off. The filtrate was evaporated and the residue was subjected to fractional distillation to yield **6a** as a colorless oil (1.1 g, 60%), bp 94-96° (400 Pa).

3-exo-Hydroxymethylbicyclo[2.2.1]hept-5-enyl-2-exo-amine (3) and 3-endo-Hydroxymethylbicyclo[2,2,1]heptyl-2-endo-amine (4).

Lithium aluminum hydride (14.0 g, 0.37 mole) was added in portions, with stirring and cooling, to dry tetrahydrofuran (700 ml). The 3-exo-aminobicyclo[2.2.1]hept-5-ene-2-exo-carboxylic acid (20.60 g, 0.133 mole) [7] or the ethyl 3-endo-aminobicyclo[2.2.1]heptane-2-endo-carboxylate (6a) (24.37 g, 0.133 mole) was added to the mixture, which was stirred and refluxed for 20 hours. After cooling to 0°, the excess lithium aluminum hydride was decomposed by adding water (30 ml) dropwise, and the mixture was stirred at room temperature until it became completely white. The precipitate was filtered off with suction and repeatedly washed with hot tetrahydrofuran and then with ethanol, and the oily residue was subjected to fractional distillation to yield 3 and 4 as colorless oils. The aminoalcohol 3 [12.2 g, 65 %, bp 96-98° (400 Pa)] became crystalline on standing at 4°.

Anal. Calcd. for C₈H₁₈NO: C, 69.03; H, 9.41; N, 10.06. Found: C, 68.81; H, 9.35; N, 10.31.

Aminoalcohol 4 had bp 90-92° (400 Pa), vield 13.5 g (72%).

Anal. Calcd. for CaH15NO: C, 68.04; H, 10.70; N, 9.92. Found: C, 67.92; H, 10.63; N, 9.84.

2-endo-Methylamino-3-endo-hydroxymethylbicyclo[2.2.1]heptane (16) and 2-exo-Methylamino-3-exo-hydroxymethylbicyclo[2.2.1]heptane (22).

Ethyl chloroformate (1.3 g, 0.012 mole) was added dropwise to a mixture of ester salt 6 or 20 (0.01 mole), sodium hydrogen carbonate (2.52 g, 0.03 mole) and water (30 ml). The mixture was stirred for 2 hours, and after 10 hours the solid 15 (2.1 g, 82%) in the case of 6 was filtered off with suction, washed with water and dried. Colorless crystals from n-hexane, mp 64-66°.

Anal. Calcd. for C13H12NO4: C, 61.15; H, 8.29; N, 5.49. Found: C, 61.36; H, 8.40; N, 5.28.

In the case of 20, the oily product 21 was extracted with chloroform, and the extract was washed with water, dried (sodium sulfate) and evaporated. The oily residue was reduced without any purification.

Lithium aluminum hydride (2.0 g) was suspended in dry tetrahydrofuran (100 ml) and the product 15 or 21 was added in portions, with stirring. The mixture was refluxed for 1 hour and the excess lithium aluminum hydride was decomposed with water (5 ml). After the usual working-up, colorless oily products were obtained: 16, bp 94-96° (400 Pa; 1.18 g, 76% calcd. for 6) and 22, bp 96-98° (400 Pa; 1.10 g, 71% calcd. for 20). For analytical purposes the hydrochlorides were prepared.

Anal. Calcd. for C₀H₁₀ClNO: C, 56.39; H, 9.46; N, 7.30. Found: (16) C, 56.44; H, 9.29; N, 7.37. Found: (22) C, 56.27; H, 9.30; N, 7.44. 2-Aryl-5,8-methano-r-4a,c-5,c-8a-tetrahydro-4H-3,1-benzoxazines (7a-d)

and 2-Aryl-5,8-methano-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazines (8a-d).

The aminoalcohol 3 or 4 (1.4 g, 0.01 mole) and the imidate [Ar = C_6H_5 : 1.5 g, Ar = C_6H_4Cl (m or p), 1.83 g, Ar = $C_6H_4CH_3(p)$, 1.63 g] were dissolved in ethanol (20 ml), one drop of ethanol saturated with hydrogen chloride was added, and the mixture was refluxed. After the reaction was complete (6-8 hours), the mixture was concentrated and the residue was crystallized from ethanol-petroleum ether. Data on the prepared colorless crystalline compounds are listed in Table I.

5.8-Methano-r-4a,c-5,c-8,c-8a-tetrahydro-4H-3,1-benzoxazin-2(1H)-one (11), 5,8-Methano-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazin-2(1H)-one (12), 5,8-Methano-1-methyl-r-4a,c-5,6,7,c-8a-hexahydro-4H-3,1benzoxazin-2(1H)-one (17) and 5,8-Methano-1-methyl-r-4a,t-5,6,7,t-8,c-8ahexahydro-4H-3,1-benzoxazin-2(1H)-one (23).

Ethyl chloroformate (1.1 g, 0.01 mole) was added dropwise to a mixture of aminoalcohol (3: 1.40 g, 4: 1.42 g, 16 and 22: 1.56 g, 0.01 mole), water (10 ml) and sodium hydrogen carbonate (0.9 g, 0.01 mole). The mixture was stirred and refluxed for 5 minutes and then concentrated. The dry residue was heated with sodium methoxide at 120° for 20 minutes in an oil bath. After cooling, the mixture was repeatedly extracted with hot ethyl acetate, the extracts were combined and the solvent was evaporated. The residue was transferred onto an aluminium oxide (neutral, activity grade 2) column, and eluted with benzene and then with chloroform. After evaporation of the chloroform, the residue gave colorless crystals (11 and 12) from ethyl acetate-petroleum ether, or colorless oils (17 and 23).

The mp's, yields and analyses of compounds 11, 12, 17 and 23 are listed in Table I.

5,8-Methano-r-4a,c-5,c-8,c-8a-tetrahydro-4H-3,1-benzoxazine-2(1H)-thione (13), 5,8-Methano-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazine-2(1H)-thione (14), 5,8-Methano-1-methyl-r-4a,c-5,6,7,c-8a-hexahydro-4H-3,1-benzoxazine-2(1H)-thione (18) and 5,8-Methano-1-methyl-r-4a,t-5,6,7,t-8,c-8a-hexahydro-4H-3,1-benzoxazine-2(1H)-thione (24).

The aminoalcohol (3: 2.30 g, 4: 2.34 g, 16 and 22: 2.57 g, 0.0165 mole) in a solution (10 ml) of potassium hydroxide (1.1 g) was cooled to 0°, carbon disulphide (1.3 g) in dioxane (8 ml) was added and the mixture was stirred for 5 minutes. Potassium hydroxide (0.55 g) in water (10 ml) and then an aqueous solution (30 ml) of lead(II) nitrate (5.5 g) were added, followed by stirring at 60° for 10 minutes. The lead sulphide was filtered off, washed with hot water and extracted with hot ethanol. The aqueous filtrate and the ethanolic extracts were combined and evaporated to dryness. The residue gave colorless crystals from ethanol.

The data on compounds 13, 14, 18 and 24 are listed in Table I.

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